organic papers

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Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.002 \text{ Å}$ R factor = 0.045 wR factor = 0.117 Data-to-parameter ratio = 13.8

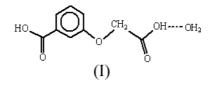
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-Carboxyphenoxyacetic acid monohydrate

The asymmetric unit of the title compound, $C_9H_{10}O_6 \cdot H_2O$, consists of one 3-carboxyphenoxyacetic acid (3-CPOAH₂) and one water molecule. The intermolecular interactions link symmetry-related molecules into hydrogen-bonded dimers. A supramolecular hydrogen-bonding network structure is formed *via* further intermolecular hydrogen bonds involving the water molecule.

Comment

Carboxyphenoxyacetic acids, which have been known to show biological activities and are widely used in agriculture, are a family of excellent ligands with versatile binding modes. To the best of our knowledge, several metal complexes with 4carboxyphenoxyacetic acid and 2-carboxyphenoxyacetic acid have been reported (Wai *et al.*, 1990; Kennard *et al.*, 1986; Gao *et al.*, 2004). However, there has been little structural information on 3-carboxyphenoxyacetic acid to date. The reaction of chloroacetic acid with 3-hydroxybenzoic acid under basic conditions yielded the title compound, (3-CPOAH₂)·H₂O, (I), whose structure is reported here.



As shown in Fig. 1, the asymmetric unit consists of a molecule of 3-carboxyphenoxyacetic acid and one water molecule, which are linked by two hydrogen bonds, one involving atom O1 and the other involving O5 (see Table 2), leading to a supramolecular network structure. The bond lengths of the carboxyl O atoms of the oxyacetic acid group are slightly shorter than those of benzoic acid. The C3-O3-C2 and O3-C2-C1 bond angles are 118.9 (1) and 107.6 (1) Å, respectively. The oxyacetic acid group and the benzene ring

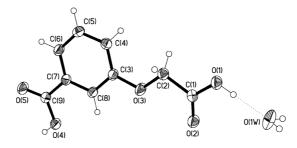


Figure 1

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved View of the title compound, with 30% probability displacement ellipsoids. The dashed line indicates a hydrogen bond.

Received 8 March 2004 Accepted 2 April 2004 Online 17 April 2004 are almost coplanar, with a mean deviation of 0.066 Å. A dimeric unit, analogous to a crown ether ring, is formed by an $O-H\cdots O$ bond between the carbonyl O atom of the oxyacetic acid group and the carboxyl O atom of an adjacent molecule (see Fig. 2 and Table 2).

Experimental

The title compound was prepared by the addition of 3-hydroxybenzoic acid (0.05 mol) to an aqueous solution of chloroacetic acid (15 mmol), the pH being adjusted to 9–10 with 0.2 *M* NaOH solution. The resulting mixture was refluxed for 4 h and then cooled to room temperature. The pH was adjusted to 2 by the addition of 0.1 *M* HCl with subsequent filtration. Colorless plate-like single crystals were obtained from the filtrate at room temperature over a period of several days. IR spectroscopic analysis: (KBr, ν cm⁻¹): 3376, 1725, 1377, 1256, 1093; analysis calculated for C₉H₁₀O₆: C 50.47, H 4.71%; found: C 50.61, H 4.89%.

Crystal data

2	
$C_9H_8O_5 \cdot H_2O$	$D_x = 1.545 \text{ Mg m}^{-3}$
$M_r = 214.17$	Mo $K\alpha$ radiation
Monoclinic, P_{2_1}/n	Cell parameters from 6464
a = 8.728 (2) Å	reflections
b = 4.847 (1) Å	$\theta = 3.9-27.4^{\circ}$
c = 21.935 (4) Å	$\mu = 0.13 \text{ mm}^{-1}$
$\beta = 97.22 \ (3)^{\circ}$	T = 293 (2) K
$V = 920.6 (3) \text{ Å}^3$	Plate, colorless
Z = 4	$0.38 \times 0.26 \times 0.11 \text{ mm}$
Data collection	
Rigaku R-AXIS RAPID	2097 independent reflections
diffractometer	1640 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.019$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(ABSCOR; Higashi, 1995)	$h = -11 \rightarrow 11$
$T_{\min} = 0.960, T_{\max} = 0.986$	$k = -6 \rightarrow 6$
3435 measured reflections	$l = -28 \rightarrow 28$
Deference out	

Refinement

Refinement on F^2	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.045$	independent and constrained
$wR(F^2) = 0.117$	refinement
S = 1.02	$w = 1/[\sigma^2 (F_o^2) + (0.0647P)^2]$
2097 reflections	where $P = (F_o^2 + 2F_c^2)/3$
152 parameters	$(\Delta/\sigma)_{\rm max} < 0.001$
	$\Delta \rho_{\rm max} = 0.27 \text{ e } \text{\AA}^{-3}$
	$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$

Table 1

Calasta J		parameters ((A O'	`
Selected	geometric	narameters i	A	
Selected	Scometile	purumeters (· · · ,	<i>,</i> .

O1-C1	1.296 (2)	O4-C9	1.319 (2)	
O2-C1	1.199 (2)	O5-C9	1.211 (2)	
O1-C1-C2	110.8 (1)	O4-C9-C7	113.3 (1)	
O2-C1-O1	125.0 (2)	O5-C9-O4	123.4 (2)	
O2-C1-C2	124.2 (2)	O5-C9-C7	123.3 (2)	
O3-C2-C1	107.6 (1)	C3-O3-C2	118.9 (1)	

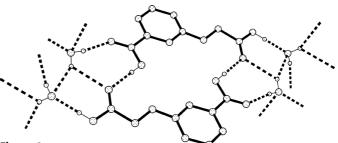


Figure 2

The hydrogen-bonded framework. The dashed lines indicate hydrogen bonds.

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O1-H10\cdots O1W$	0.86 (2)	1.71 (2)	2.563 (2)	179 (3)
$O4-H11\cdots O2^{i}$	0.84(2)	1.93 (2)	2.765 (2)	179 (2)
$O1W-H12A\cdots O2^{ii}$	0.82(2)	2.43 (3)	2.967 (2)	124 (2)
O1W-H12 A ···O1 W ⁱⁱ	0.82(2)	2.34 (2)	3.057 (2)	147 (2)
$O1W-H12B\cdots O5^{iii}$	0.84 (2)	1.91 (2)	2.740 (2)	169 (3)

Symmetry codes: (i) 2 - x, 3 - y, -z; (ii) $\frac{3}{2} - x$, $y - \frac{1}{2}$, $\frac{1}{2} - z$; (iii) $x - \frac{1}{2}$, $\frac{5}{2} - y$, $\frac{1}{2} + z$.

The H atoms attached to C were placed in calculated positions, with C-H = 0.93 or 0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$, and were included in the refinement in the riding-model approximation. The H atoms of the water molecule and carboxyl groups were located in Fourier difference maps and refined isotropically, with the O-H distances restrained to 0.82 (1) Å.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2002); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997*a*); molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* (Sheldrick, 1997*b*).

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